STEREOSELECTIVITY OF CYCLOGLYCOSYLATION IN MANNOOLIGOSE SERIES DEPENDS ON CARBOHYDRATE CHAIN LENGTH: SYNTHESES OF MANNO ISOMERS OF β - AND γ - CYCLODEXTRINS

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Abstract: Syntheses of manno isomer of β - and γ -cyclodextrins were achieved for the first time employing methylthioglycosides of α -(1 \rightarrow 4) linked mannoheptaose and mannooctaose.

Recently we reported¹ an efficient and stereoselective cycloglycosylation of thioglycoside 1 into completely benzylated cyclomannohexaose 2 in 92% yield which was subsequently deprotected into manno isomer 3 of α -cyclodextrin. In continuation of our project on cyclooligoglycose synthesis, we report here first syntheses of manno isomers (6 and 11) of β and γ -cyclodextrins.



Cycloglycosylation of 4 in the presence of $PhSeOTf^2$ in $(ClCH_2)_2$ at -22° afforded 5³ and 7 in 46 and 33% yield which were then hydrogenolysed in the presence of 20% Pd(OH)₂/C in 12:1:1 MeOH-EtOAc-H₂O to give 6 and 8, respectively. Similarly, cycloglycosylation of 9 afforded 10 and 12 in 53 and 25% yield which were again quantitatively deprotected to 11 and 13, respectively.

The structures of the α - $(1\rightarrow 4)$ linked product 5 and the β - $(1\rightarrow 4)$ linked product 7 were deduced from ¹H and ¹³C-n.m.r. data measured in CDCl₃. A signal for H-1 and C-1 of 5 was observed, respectively, at δ 5.033 as a singlet and at δ 100.8 with a value of ¹J_{C,H} 164 Hz, in agreement with C₇ symmetry of the molecule. For the compound 7, we observed signals for seven anomeric protons at δ 5.283 (s, 1H), 5.061 (d, 1.5 Hz, 2H), 5.050 (s, 1H), 5.012 (d, 1.5 Hz, 1H), 4.990 (d, 1.5 Hz, 1H) and 4.574 (s, 1H). Among these signals for anomeric protons, a signal at δ 4.574 could be assigned for H-1¹ with β -D configuration by ¹H-¹³C heteronuclear multiple quantum coherence (HMQC)⁴ which revealed a corresponding signal for C-1¹ at δ 96.67 with ¹J_{C,H} 154 Hz⁵ as well as the signals for C-1^{2~7} with α -D configuration at δ 101.86, 101.21, 100.94, 100.87, 100.83 and 99.08 (¹J_{C,H} ~170 Hz⁵). The assignment of configuration for the newly introduced glycosidic linkage in 10 and 12 was also made by ¹H-¹³C HMQC measurements. These experimental results showed that cycloglycosylations of α -(1 \rightarrow 4)-linked thio-mannooligosides 1, 4, and 9 in the presence of PhSeOTf afforded a high yield (80~90%) of cyclization products but the





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stereoselectivity of cycloglycosylation depends heavily on the chain length of the oligosaccharides employed. The α/β ratios of the products were as follows: only α for 1, 1.4/1 for 4, and 2.1/1 for 9.

The key intermediates 4 and 9 were prepared starting from the protected mannobiosyl derivatives 14^1 and 15^1 . Conversion of 14 into trichloroacetimidate 16 was performed in two steps (1 CAN in 4:1 MeCN-H₂O⁶, 2 CCl₃CN⁷, DBU in (CH₂Cl)₂, overall 74%). TMSOTf⁸ promoted glycosylation of 15 with 16 in (ClCH₂)₂ at -20° afforded the mannotetraosyl derivative 17 in 85% yield. Conversion of 17 into a glycosyl donor 18 was accomplished in two steps (overall 64%) as in the case of 14. Treatment of 18 with Bu₃SnSMe⁹ in the presence of BF₃·OEt₂ gave an 88% yield of 19 which was converted into a glycosyl acceptor 20 in 95% yield by treatment with (NH₂)₂CS in EtOH¹⁰.



Scheme 2 (CA = CICH₂CO, MBZ = 4-MeBz, MP = 4-MeOPh)

Having prepared necessary mannotetraosyl donor 18 and acceptor 20, glycosylation of 20 with the known mannotriosyl donor 21^{1} in the presence of TMSOTf afforded a 66% yield of mannoheptaosyl derivative 22 which was then converted into a key intermediate 4 in 5 steps (1 (NH₂)₂CS in EtOH, 2 EtOCH=CH₂, PPTS in (CICH₂)₂, 3 NaOMe in 1:3 THF-MeOH, 4 BnBr, NaH in DMF, 5 Amberlyst 15 (H⁺) in 1:1 CH₂Cl₂-MeOH, overall 64%). Similarly, glycosylation of 20 with mannotetraosyl donor 18 in the presence of TMSOTf gave a 62% yield of mannooctaosyl derivative 23 which was then converted in 5 steps into another key intermediate 9 in 57% overall yield in a same reaction sequence as described for 4.

Crucial cycloglycosylations of 4 and 9 were achieved in high yield but with low stereocontrol as described already. Reasonable explanation for these interesting experimental observations has remained to be done.

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Reference and Notes

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- 3) Physical data for key compounds are described below. Values of $[\alpha]_D$ and $\delta_{H,C}$ were measured for CHCl₃ and CDCl₃ solutions, respectively, at $23\pm3^\circ$, unless noted otherwise. 4: $[\alpha]_D$ -1.3° (c 0.8); SH 5.327 (d, 3H, 2.1 Hz), 5.300 (s, 2H) and 5.284 (s, 2H) for 7 x H-1, 2.169 (s, SMe); SC 99.9 $(^{1}J_{CH} 171 Hz, 2 \times C^{-1}), 99.7$ $(^{1}J_{CH} 173 Hz, 4 \times C^{-1}), 82.7$ $(C^{-1}), 13.8$ (SMe). 5: $[\alpha]_{D}$ -36.7° (c 0.5). 6: [α] +47.4° (c 0.3, H₂O); δH (D₂O, 50°) 4.941 (d, 2.1 Hz, H-1), 4.039 (dd, 2.1 and 3.4 Hz, H-2), 3.924 (dd, 3.4 and 8.9 Hz, H-3), 3.897 (dd, 2.4 and 12.5 Hz, H-6), 3.857 (dd, 4.6 and 12.5 Hz, H-6), 3.792 (ddd, 2.4, 4.6 and 8.9 Hz, H-5), 3.736 (t, 8.9 Hz, H-4). 7: [a]D -29.4° (c 0.5). 8: [a]D +50.0° (c 0.3. H2O); 8H (D2O, 50°) 5.195 (d, 2.4 Hz), 5.005 (d, 2.6 Hz), 4.987 (d, 1.8 Hz), 4.962 (d, 2.0 Hz), 4.950 (d, 2.1 Hz), and 4.934 (d, 2.1 Hz) for 6 x H-1, 4.930 (s, H-1¹), 9: $[\alpha]_{D}$ -1.6° (c 0.3); δ H 5.330 (s, 4H), 5.299 (s, 2H) and 5.281 (s, 2H) for 8 x H-1, 2.169 (s, SMe); SC 99.9 (¹JCH 171 Hz, 2 x C-1), 99.7 $(^{1}J_{CH} 170 \text{ Hz}, 5 \text{ x C-1}), 82.7 (C-1^{1}), 13.8 (SMe). 10: [a]D -40.7^{\circ} (c 1.7); \delta_{H} 5.073 (d, 1.5 \text{ Hz}, H-1).$ 11: [a] +27.2° (c 0.3, H2O); δ H (D2O, 50°) 4.981 (d, 2.1 Hz, H-1). 12: [a] -42.9° (c 1.0); δ H (HMQC) 5.316, 5.255, 5.163, 5.073 (2H), 5.051 and 5.042 (6s, 7 x H-1), 4.887 (s, H-1¹); $\delta_{\rm C}$ (HMQC) 97.4 (¹JCH 157 Hz, C-1¹), 100.4~100.8(¹JCH ~170 Hz, 7 x C-1). 13: [a]D +61.9° (c 0.2, H₂O); δH (D₂O, 50°) 5.185 (d, 2.1 Hz), 5.105 (d, 1.8 Hz), 5.084 (d, 2.1 Hz), 5.063 (d, 1.8 Hz), 4.997 (d, 1.8 Hz), 4.989 (d, 2.1 Hz) and 4.975 (d, 1.8 Hz) for 7 x H-1, 4.950 (s, H-1¹). 16: $[\alpha]_D$ -16.3° (c 0.9); δ_H 8.743 (s, C=NH), 6.404 (d, 1.8 Hz, H-1¹), 5.719 and 5.702 (2dd, 1.8 and 3.1 Hz, H-2^{1,2}), 5.608 (d, 1.8 Hz, H-1²), 2.385 and 2.376 (2s, 2 x PhMe). 17: [a]D -33.5° (c 4.4); SH 5.597 (d, 1.8 Hz, 2 x H-1), 5.578 (d, 1.5 Hz, H-1), 5.528 (d, 1.5 Hz, H-1), 5.470 (t, 10.0 Hz, H-4⁴); $\delta_{\rm C}$ 99.3 (C-1), 98.9 (2 x C-1), 96.9 (C-1¹), 55.5 (OMe). 18: $[\alpha]_D$ -30.6° (c 0.9); δ_H 8.778 (s, C=NH), 6.430 (d, 2.0 Hz, H-1¹), 5.584 (s, 2 x H-1), 5,530 (d. 1.8 Hz, H-1), 5,476 (t, 10.0 Hz, H-4⁴), 19: [a]D -34.0° (c 1.5); 8H 5.591 (d, 1.8 Hz, H-1), 5.529 (s, 2 x H-1), 5.477 (t, 10.0 Hz, H-4⁴), 5.333 (d, 2.0 Hz, H-1¹); & 99.3 (C-1), 98.9 (2 x C-1), 83.8 $(C-1^{1})$. 20: $[\alpha]_{D}$ -40.4° (c 1.0); δ_{H} 5.579 (d, 1.8 Hz, H-1), 5.528 (s, 2 x H-1), 5.333 (d, 1.8 Hz, H-1¹); $\delta_{\rm C}$ 99.3 (3 x C-1), 83.8 (C-1¹). 22: [a]D -42.1° (c 0.8); $\delta_{\rm H}$ 5.596 (d, 1.5 Hz, H-1), 5.584 (s, 2 x H-1), 5.563 (d, 1.5 Hz, H-1), 5.560 (d, 1.8 Hz, H-1), 5.543 (s, H-1), 5.473 (t, 10.0 Hz, H-4⁷), 5.331 (s, H-1¹); δ_{C} 98.8 (6 x C-1), 83.9 (C-1¹). 23: [α]_D -50.4° (c 0.8); δ_{H} 5.595 (2H), 5.579 (2H), 5.561, 5.553, 5.541, $(5d, 1.5 \sim 1.8 \text{ Hz}, 7 \text{ x H-1}), 5.472 \text{ (t, 10.0 Hz, H-48}), 5.329 \text{ (d, 1.2 Hz, H-11}).$
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